

19. (twice amended) A glass composition, comprising:

b1

|                                |                         |
|--------------------------------|-------------------------|
| SiO <sub>2</sub>               | 70 to 75 weight percent |
| Na <sub>2</sub> O              | 12 to 15 weight percent |
| K <sub>2</sub> O               | 0 to 5 weight percent   |
| CaO                            | > 9 weight percent      |
| MgO                            | < 4 weight percent      |
| Al <sub>2</sub> O <sub>3</sub> | 0 to 2 weight percent   |
| SO <sub>3</sub>                | 0 to 1 weight percent   |
| Fe <sub>2</sub> O <sub>3</sub> | 0 to 2 weight percent   |

wherein:

|   |  |
|---|--|
| SiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub> | ≥ 70 weight percent                    |
| Na <sub>2</sub> O + K <sub>2</sub> O              | 10 to 15 weight percent                |
| CaO + MgO   | 12.74 to less than 13.4 weight percent |
| CaO/MgO   | 2 to 5                                 |

wherein the glass composition has a log 2 viscosity in the range of about 2570°F to about 2590°F (1410°C to 1421°C) and a log 4 viscosity in the range of about 1850°F to about 1894°F (1010°C to 1034°C).

b2

24. (amended) The composition according to claim 21, wherein the melting point of the glass composition from the log 2 viscosity reduces fuel usage in preparing the glass and the bending and annealing temperatures of the glass from the log 7.6 viscosity in the range of about 1300°F to about 1350°F (704°C to 732°C) and a log 13 viscosity in the range of about 1016°F to about 1020°F (547°C to 549°C) are in the range for a higher melting glass.

25. The composition according to claim 19, wherein the ratio of CaO to MgO is 2.77 to 5.